

An efficient two-step synthesis of 3-allylindoles

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Abstract—A two-step synthetic sequence for an efficient synthesis of 3-allylindoles is described.
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1. Introduction and discussion

Several methods have been developed for the synthesis of indoles and substituted indoles as they are a key feature of many natural products and pharmaceutical compounds.^{1–20} However, there is still a need to develop newer methods for their synthesis. Alkylation of indoles or salts of indoles under drastic conditions yields a complex mixture of 1-substituted, 1,3-disubstituted, 3-substituted indoles^{21–24} and/or indolenines.²⁵ The conversion of indoles to 3-allylindoles was effected via zinc-mediated Barbier reactions²⁶ or palladium catalysed allylation.²⁷ In yet another method, acid-catalysed Claisen rearrangement of 1-allylindole²⁸ furnished 3-allylindole. All the above methods utilise indoles as starting material.^{28–32}

We herein describe an application of the Wittig olefination–Claisen rearrangement methodology³³ towards a short and efficient synthesis of 3-allylindoles in just two steps using simple starting materials viz. *o*-nitrobenzaldehydes. Wittig olefination of *o*-nitrobenzaldehyde **1a**, under standard conditions,³³ with allyloxymethylenetriphenylphosphorane furnished the corresponding allyl vinyl ether **2a**. The allyl vinyl ether **2a** was then subjected to a Claisen rearrangement by heating in refluxing diglyme–water solvent system (1:1) in the presence of excess FeSO₄ and NH₃ (**Scheme 1**). To our pleasant surprise, 3-allylindole **3a** was obtained from the reaction mixture.

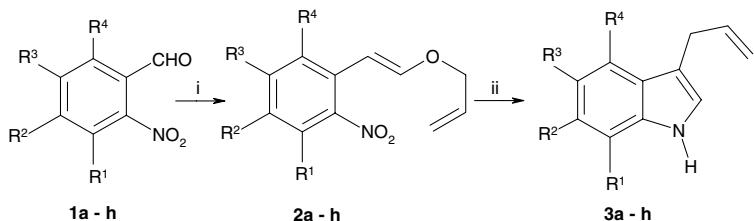
This clearly indicated that the two reactions, namely, Claisen rearrangement of the allyl vinyl ether to the corresponding 4-pentenal and reduction of the nitro group with FeSO₄–NH₃ to the corresponding amine were effected simultaneously. The resulting amino aldehyde underwent in situ cyclisation to furnish 3-allylindole. Thus, the synthesis of 3-allylindole was achieved in only two steps. Reaction of substituted *o*-nitrobenzaldehydes also furnished, under the above reaction conditions, substituted 3-allylindoles in good yields.

Furthermore, reaction of the substituted *o*-nitrobenzaldehydes with potassium tertiary butoxide and another Wittig salt, namely, crotyloxymethylenetriphenylphosphonium chloride, gave the corresponding allyl vinyl ethers **4** in good yields (**Scheme 2**). These allyl vinyl ethers, under the above reaction conditions, furnished novel 3-(3-but enyl)indoles **5**. The present protocol offers a general method for the synthesis of 3-allylindoles in two simple steps. The utility of these 3-allylindoles in the synthesis of complex natural products is being actively investigated.

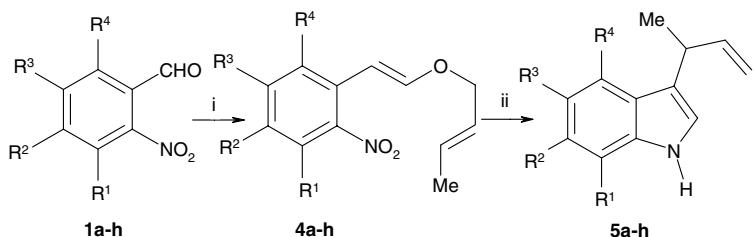
2. Experimental procedure for Claisen rearrangement/ring closure (**3a**)

To a stirred suspension of ferrous sulfate (16 g, 12 mmol) in diglyme–water (1:1, 5:5 ml), the allyl vinyl ether **2a** (mixture of *E* and *Z* isomers) (1 g, 1 mmol) was added. After refluxing the mixture for 15–20 min (TLC check), excess ammonia was added. The resulting solution was refluxed for 24 h (TLC check). The reaction mixture as such was purified by silica gel column

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	R ¹	R ²	R ³	R ⁴	% Yield ^a 2	% Yield ^a 3
a	H	H	H	H	75	77
b	OMe	H	H	H	72	80
c	H	H	H	Cl	73	79
d	H	H	Cl	H	70	76
e	H	OMe	OMe	H	70	75
f	Br	OMe	OMe	H	71	74
g	H	H	OH	H	70	76
h	H	-OCH ₂ O-		H	72	76

^a Isolated yields**Scheme 1.** Reagents and conditions: (i) CH₂=CHCH₂OCH₂P⁺Ph₃Cl⁻, t-BuO⁻K⁺, THF, 0 °C, 30 min; (ii) FeSO₄, NH₃, diglyme–water (1:1), reflux.

	R ¹	R ²	R ³	R ⁴	% Yield ^a 4	% Yield ^a 5
a	H	H	H	H	70	75
b	OMe	H	H	H	70	74
c	H	H	H	Cl	71	74
d	H	H	Cl	H	73	78
e	H	OMe	OMe	H	72	76
f	Br	OMe	OMe	H	73	79
g	H	H	OH	H	75	78
h	H	-OCH ₂ O-		H	73	75

^a Isolated yields**Scheme 2.** Reagents and conditions: (i) MeCH=CHCH₂OCH₂P⁺Ph₃Cl⁻, t-BuO⁻K⁺, THF, 0 °C, 30 min; (ii) FeSO₄, NH₃, diglyme–water (1:1), reflux.

chromatography using the hexane–ethyl acetate (98:2) solvent system.

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References and notes

- Gribble, G. W. *J. Chem. Soc., Perkin Trans. I* **2000**, 1045–1075.
- Dalpozzo, R.; Bartoli, G. *Curr. Org. Chem.* **2005**, 9, 163–178.

3. Cheng, X.; Li, B.-d.; Lu, C.-x. *Jiangsu Huagong* **2005**, 33, 141–143.
4. (a) Shi, L.; Huo, X.; Liu, C.; Cai, T. *Cuihua Xuebao* **2005**, 26, 403–406; (b) Shi, L.; Huo, X.; Ren, S.; Xu, J. *Cuihua Xuebao* **2005**, 26, 449–450; (c) Shi, L.; Zhou, X.; Cai, T.; Wang, X. *Cuihua Xuebao* **2003**, 24, 471–474; (d) Shi, L.; Sun, J.; Ren, S.; Cai, T.; Wang, X. *Cuihua Xuebao* **2004**, 25, 384–386.
5. Schmidt, A. M.; Eilbracht, P. *Org. Biomol. Chem.* **2005**, 3, 2333–2343.
6. Campanati, M.; Franceschini, S.; Piccolo, O.; Vaccari, A. *J. Catal.* **2005**, 232, 1–9.
7. Ackermann, L.; Born, R. *Tetrahedron Lett.* **2004**, 45, 9541–9544.
8. (a) Shi, L.; Liu, C.; Song, W.; Cai, T.; Wang, X. *Fenzi Cuihua* **2004**, 18, 257–260; (b) Sun, J.; Shi, L.; Wang, X.; Su, X.; Cai, T. *Fenzi Cuihua* **2002**, 16, 229–233.
9. Gao, N.; Fu, B.; Zheng, W.; Wang, G.; Wang, Q. *Huaxue Gongye Yu Gongcheng (Tianjin, China)* **2003**, 20, 310–312.
10. Guo, C.; Wang, J.; Huang, J. *Huagong Xuebao (Chinese Edition)* **2004**, 55, 1201–1205.
11. Shi, L.; Sun, J. M.; Wang, X. P.; Su, X. Y.; Cai, T. X. *Chin. Chem. Lett.* **2002**, 13, 211–212.
12. Arisawa, M.; Nishida, A. *Kagaku Kogyo* **2004**, 55, 120–123.
13. Tokuyama, H. *Yakugaku Zasshi* **2003**, 123, 1007–1021.
14. Sekar, N. *Colourage* **2003**, 50, 65–66.
15. Shi, L.; Song, W.; Xue, J. *Huaxue Tongbao* **2003**, 66, w058/1–w058/6.
16. Koehling, P.; Schmidt, A. M.; Eilbracht, P. *Org. Lett.* **2003**, 5, 3213–3216.
17. Wang, X.; Zhuang, Q.; Wang, X.; Shi, D. *Ziran Kexueban* **2002**, 20, 62–64.
18. Hiroi, K.; Hiratsuka, Y.; Watanabe, K.; Abe, I.; Kato, F.; Hiroi, M. *Tetrahedron: Asymmetry* **2002**, 13, 1351–1353.
19. Downing, R. S.; Kunkeler, P. J. *Fine Chem. Through Heterogeneous Catal.* **2001**, 178–183.
20. Yamazaki, K.; Kondo, Y. *J. Comb. Chem.* **2002**, 4, 191–192.
21. Robert, M. M.; Ku, Y. Y.; Tuncay, M. S. *Tetrahedron Lett.* **1987**, 28, 3071–3074.
22. Ernest, W. E.; Charles, A.; Vitor, F. F.; Enrique, L. M.; Serge, R. P.; Jyh, H. S.; Charles, S. S. *J. Org. Chem.* **1986**, 51, 2343–2351.
23. Rubottom, G. M.; Chabala, J. C. *Synthesis* **1972**, 566–567.
24. Bocchi, V.; Casnati, G.; Dossena, A.; Villani, F. *Synthesis* **1976**, 414–416.
25. Cardillo, B.; Casnati, G.; Pochini, A.; Ricca, A. *Tetrahedron* **1967**, 23, 3771–3783.
26. Yadav, J. S.; Reddy, B. V. S.; Muralikrishna Reddy, P.; Srinivas, Ch. *Tetrahedron Lett.* **2002**, 43, 5185–5187.
27. Billups, W. E.; Erkes, R. S.; Reed, L. E. *Synth. Commun.* **1980**, 10, 147–154.
28. Inada, S.; Nagai, K.; Takayanagi, Y.; Okazaki, M. *Bull. Chem. Soc. Jpn.* **1976**, 49, 833–834.
29. Bommeijin, S.; Martin, C. G.; Kennedy, A. R.; Lizos, D.; Murphy, J. A. *Org. Lett.* **2001**, 3, 3405–3407.
30. Kamijo, S.; Yamamoto, Y. *J. Am. Chem. Soc.* **2002**, 124, 11940–11945.
31. Fleming, I.; Woolias, M. *J. Chem. Soc., Perkin Trans. 1* **1979**, 829–837.
32. Ito, Y.; Kobayashi, K.; Seko, N.; Saegusa, T. *Chem. Lett.* **1979**, 1273–1276; Ito, Y.; Kobayashi, K.; Seko, N.; Saegusa, T. *Bull. Chem. Soc. Jpn.* **1984**, 57, 73–84.
33. Kulkarni, M. G.; Pendharkar, D. S.; Rasne, R. M. *Tetrahedron Lett.* **1997**, 38, 1459–1462.